



# **CHEMISTRY**

# **BOOKS - SHARAM PUBLICATION**

# **SOLVED PAPER 4**

## Exercise

**1.** The bleaching action of  $Cl_2$  is due to ..... and that due to  $SO_2$  is

due to .....



**2.** Write the iUPAC name of  $(NH_4)_3 [Cr(SCN)_6]$ 

| <b>3.</b> What is mean by pyrometallurgy ?  |
|---|
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|   |
| <b>4.</b> What is demulsification?How is it achieved?   |
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|   |
| 5. Define space lattice.  |
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|   |
| 6. Fill up the blank in the following reaction . $C_6H_5CONH_2+Br_2+4KOH	o\ldots\ldots+2KBr+K_2CO_3+2H_2CO_3$ |
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7. Name two bread spectrum antiblotics.

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|   |
|   |
| <b>8.</b> Carbyl amine reaction is given by |
|   |
| A. $2^\circ$ amine                          |
| B. $1^{\circ}$ amine                        |
| C. $3^{\circ}$ amine                        |
| C. 5 annie                                  |
| D. Both $1^\circ~{ m and}~2^\circ$ amine    |
|   |
| Answer:                                     |
|   |
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|   |

9. Which of the following solutions has highest depression in freezing

point at one atmospheric pressure?

A. 0.1 MNaCl

 $B. 0.1 MBaCl_2$ 

 ${\rm C.}\, 0.1 MUrea$ 

 $D.\,0.1MFeCl_3$ 

Answer:

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**10.** Which of the following statements is incorrect about the collision theory of chemical reactions?

A. In considers reacting molecules or atoms to be hard sphere and

ignores their structural features

B. Number of effective collisions determines the rate of reaction

C. Collision of atoms or molecules possesing sufficient threshold

energy results into product formation

D. Molecules should collide with sufficient threshold energy and

proper orientation for the collision to be effective

### Answer:

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**11.** Which of the following is not applicable to the phenomenon of adsorption?

A.  $\Delta H > 0$ 

B.  $\Delta G < 0$ 

C.  $\Delta S < 0$ 

D.  $\Delta H < 0$ 

### Answer:

**12.** In the reaction  $NaCl + K_2Cr_2O_7 + Conc. H_2SO_4 \xrightarrow{heat} X + Na_2SO_4 + K_SO_4 + H_2O$ , 'X' is a reddish brown gas which gives a yellow solution on passing through water and a yellow pracipitate on treating the solution with lead acetate solution. So 'X' could be

A.  $Cl_2$ 

B.  $Cr_2O_3$ 

 $\mathsf{C}.\,H_2CrO_4$ 

 $\mathsf{D.}\, CrO_2 Cl_2$ 

## Answer:

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13. The most suitable reagent for the conversion of

 $RCH_2OH 
ightarrow RCHO$  is

A. PCC

B.  $KMnO_4$ 

 $\mathsf{C.}\,K_2Cr_2O_7$ 

D.  $CrO_3$ 

Answer:

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14. Which of the following polymer contains nitrogen?

A. Polyvinyl chloride

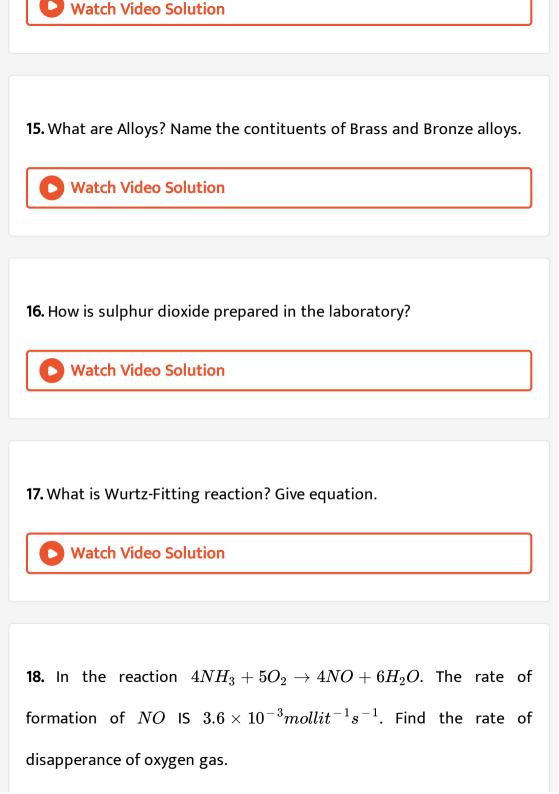
B. Bakelite

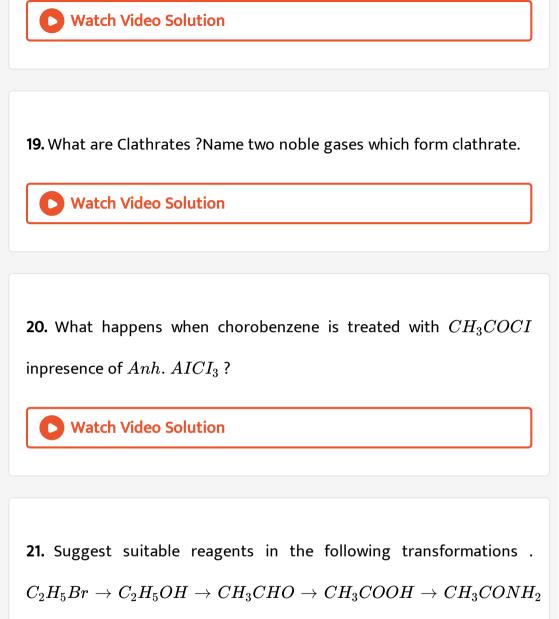
C. Nylon

D. Terylene

Answer:







**22.** The coagulation of 100 ml of a colloidal solution of gold is completely prevented by addition of 0.25g of starch to it before adding 10 ml of 10% *NaCl* solution. Calculate the gold number of starch.

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|---|--------|-------|------|-------|
|   | Tracen | Thate | 5010 |       |

**23.** Write a note on  $S_N 1$  reaction.

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24. Write a note on Hofmann ammonolysis of alkylhalides.



25. Write notes on Gabriel phthalimide synthesis.

26. What do you meant by activity and selectivity of the catalyst?

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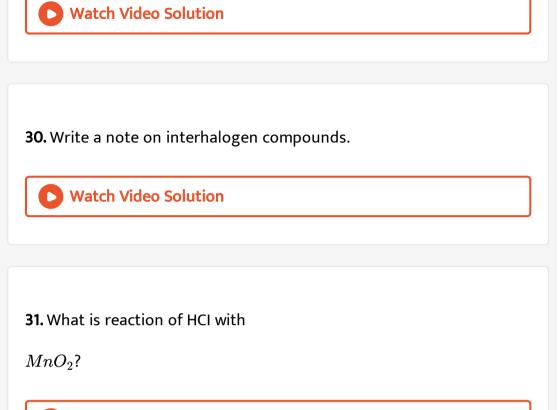
**27.** The freezing point depression of 0.1 molal solution of acetic acid in benzene is  $0.256 K kgmol^{-1}$ . Kf for benzene is  $5.12 K kg Mol^{-1}$ . What conclusion can you draw about the molecular state of acetic acid in benzene.

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28. Discuss the construction and working of a dry cell.

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**29.** How will you convert nitrobenzene to benzoic acid?



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32. What is reaction of HCI with

Ozone?

**33.** Write a note on haloform reaction.



34. State and explain Kohlrausch's law of independeat migration of

ions.

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## 35. Write the Nernst equation and emf of the following cell at 298K.

$$Fe(s)ig|Fe^{2\,+}\,(0.001M)ig|ig|H^{\,+}\,(1M)ig|H_2(g)(1\mathrm{bar})\,\mid\,Pt.$$



**36.** How can you prepare benzaldehyde from toluene?What is the

name of the reaction?





**37.** Give a chemical test to distinguish between Formaldehyde and Acetone.

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**38.** How is  $SO_2$  prepared in laboratary ?How does it react with

acidified  $KMnO_4$  solution.

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**39.** How does  $SO_2$  react with

lime water?

40. How is sulphur dioxide prepared in the laboratory ? How does it

react with

hot Mg ribbon

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**41.** What are addition and condensation polymerisation? Give one example of each.

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**42.** What is biodegradable polymer ? Give an example.



43. How is Dacron synthesised?



**44.** What do you mean by elevation boiling point ?How will you calculate the molecular mass of a solute from elevation of boiling point?

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**45.** 18g of glucose  $(C_6H_{12}O_6)$  is dissolved in 1kg of water. At what temperature will solution boil at 1 atmosphere?  $K_b$  for water is  $0.52Kkgmol^{-1}$ .