

## **CHEMISTRY**

# **BOOKS - ARIHANT PUBLICATION JHARKHAND**

## **SOLVED PAPER 2018**

## **Section Ii Chemistry**

**1.** Kurkure packet's empty space is filled with an inert gas, in order to prevent it from

A. reduction

B. oxidation

- C. oxidation-reduction
- D. precipitation

#### **Answer: B**



- **2.** Transparent photographic films gradually change their colour when exposed to light due to.
  - A. combination reaction
  - B. thermal decomposition
  - C. oxidation reaction
  - D. displacement reaction

## **Answer: B**



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- **3.** Aluminium when exposed to  $O_2$  atmosphere, formation of  $Al_2,\,O_3$ , is observed. It is.
  - A. corrosion
  - B. rusting
  - C. decomposition
  - D. rancidity

## **Answer: A**



**4.** When blue coloured  $CuSO_4$ , crystals is heated, it gives a white amorphous solid. This white solid is

A. 
$$CuSO_4$$
.  $H_2O$ 

- B.  $CuSO_4$
- C. Cuo
- D.  $CuCO_3$

#### **Answer: B**



A. pink
B. yellow
C. colour less
D. orange
Answer: C
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<b>6.</b> Second highest electronegative element in periodic
table

5. Colour of phenolphthalein in acidic medium

A. fluorine B. chlorine C. oxygen D. nitrogen **Answer: C Watch Video Solution** 7. The element with half-filled outermost electronic configuration A. aluminium B. magnesium

- C. boron
- D. nitrogen

#### **Answer: D**



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# **8.** The order of effective nuclear charge is

A. 
$$LI < Na < K < Rb$$

$$\operatorname{B.}LI>Na>K>Rb$$

$$\mathsf{C.}\,LI < K < Na < Rb$$

D. 
$$LI < Rb < K < Na$$

### **Answer: B**



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# **9.** Triad among the following is

A. H, F, CI

B. F, CI, Br

C. Be, Mg, Ca

D. Ca, Sr, Ba

## **Answer: C**



## 10. KOH is stronger base than LiOH. The reason is

- A. Li is highly reactive
- B. atomic size of Li is less
- C. K is highly reactive
- D. atomic size of K is more

#### **Answer: C**



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11. Hydrolysis of an ester in presence of alkali is called

A. esterification

- B. saponification
- C. substitution
- D. addition

## **Answer: B**



- 12. Vegetable oils usually contain
  - A. unsaturated fatty acids
  - B. carboxylic acids
  - C. saturated fatty acids
  - D. glacial acetic acid

## **Answer: A**



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13. Hard water forms scum with soap, due to the presence of

A.  $CH_3$ , COOH

B. NaOH

 $\mathsf{C.}\,\mathit{Ca}(OH)_2$ 

D.  $CaCl_2$ 

#### **Answer: D**



**14.** Hydrophobic end of micelle is solube in hydrocarbon because it is

- A. non-polar
- B. polar
- C. ionic
- D. covalent

**Answer: A** 



**15.** Compound which does not belong to the homologous series

- A.  $CH_4O$
- B.  $C_4H_{12}O$
- $\mathsf{C}.\,C_2H_6O$
- D.  $C_3H_8O$

**Answer: B** 



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16. Colour of Taj Mahal in Agra is changed due to

A. sunlight B. acid rain C. slaked lime D. corrosion **Answer: B Watch Video Solution** 17. Antacids are used to remove A. body pain B. acidlty C. basicity

D. headache

#### **Answer: B**



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18. Chemical compound present in tooth paste is.

A. basic- $Ca_3(PO_4)_2$ 

B. acidic- $Ca_3(PO_4)$ 

C. basic-Ca  $(HCO_3)_2$ 

D. salt-  $Ca_3P_2$ 

**Answer: A** 

# **19.** pH of orange juice is

- A. 2.00-2.35
- B. 3.12-3.33
- C. 3.00-3.75
- D. 3.3-4.19

#### **Answer: B**



B. hydrochloric acid C. acetic acid D. citric acid **Answer: C Watch Video Solution** 21. Allotrope of carbon which posses high melting point? A. Graphite B. Fullerene

A. amino acid

- C. Coke
- D. Diamond

### **Answer: D**



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# **22.** The compound gives a sooty flame on burning .

- A.  $CH_4$
- B.  $C_3H_3$
- $\mathsf{C}.\,C_2H_6$
- D.  $C_2H_4$

### **Answer: D**



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**23.** Addition of hydrogen to alkene in presence of nickel catalyst is

- A. reduction
- B. oxidation
- C. addition and elimination
- D. substitution

## **Answer: C**



**24.** A compound 'A' with formula  $C_3H_8O$  possess two structural isomers. On oxidation it produces an acid. Compound A is

- A. 1-propanal
- B. 2-propanone
- C. 1-propanol
- D. 2-propanal

## **Answer: C**



**25.** Poisonous organic compound present in denatured rectified spirit is

- A. ethyl alcohol
- B. methyl alcohol
- C. ethanolo acid
- D. formic acid

#### **Answer: A**



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**26.** Method used to prevent the corrosion of aluminium.

A. galvanisation B. anodising C. tanning D. cathodic protection **Answer: B Watch Video Solution 27.** Metal A' reacts with dil.  $HNO_3$  and liberates hydrogen. A is A. Mg B. Ag

- C. Cu
- D. Au

## **Answer: A**



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# 28. The metal which is highly reactive

- A. Hg
- B. Zn
- C. Pb
- D. Al

### **Answer: D**



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- 29. Among the following the polar compound is .
  - A. kerosene
  - B. alcohol
  - C. benzene
  - D. gasoline

#### **Answer: B**



**30.** Metals can be extracted easily and economically from.

A. mineral

B. metallic compound

C. ore

D. rock

#### **Answer: C**



**31.** The process of formation of covalent bonds with other atoms to form long chains is called

- A. addition
- B. coordination
- C. catenation
- D. polymerisation

#### **Answer: C**



32.

$$CH_3COOH + C_2H_5OH \stackrel{X}{\longrightarrow} CH_3COOC_2H_5 + H_2O, X$$

is

A.  $H_2SO_4$ 

B. NaOH

C.  $KMnO_4$ 

D. Ni

## **Answer: A**



## 33. Modern periodic table is based on

- A. atomic mass
- B. mass number
- C. atomic number
- D. electronic configuration

## **Answer: C**



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**34.** In Mendeleev's periodic table elements with atomic numbers 18, 35 belongs to which group?

A. Group zero, I B. Group zero, III C. Group zero, D. Group zero, VII **Answer: D Watch Video Solution 35.** The element which is electropositive in nature? A. Sulphur B. Silicon C. Boron

D. Copper

#### **Answer: C**



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**36.** Burning of Mg in air is.

A. physical change

B. chemical change

C. sublimation

D. oxidation-reduction

**Answer: B** 

**37.** Reaction of marble with atmospheric humidity and  $CO_2$  to form slaked lime is an example for .

- A. decomposition rection
- B. displacement reaction
- C. combinaiton reaction
- D. redox reaction

**Answer: C** 



**38.** A compound X reaction with  $CO_2$  present in air to from Y . Y one decomposition produces on oxide . The compound X is .

- A.  $CaCO_3$
- B. CaO
- C.  $Ca(HCO_3)_2$
- D.  $Ca(OH)_2$

**Answer: B** 



**39.** Barium can displace which element from its sulphate?

- A. Sodium
- B. Caesium
- C. Aluminium
- D. Hydrogen

#### **Answer: A**



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**40.** Dissolution of glucose is

A. endothermic reaction B. exothermic reaction C. redox reaction D. displacement reaction **Answer: C Watch Video Solution 41.** Water of crystallisation is not present in A. washing soda B. baking soda

C. soda line

D. gypsum

#### **Answer: B**



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## **42.** Plaster of Pairs is

A. 
$$CaSO_4.2H_2O$$

B. 
$$CaSO_4$$
.  $\frac{1}{2}H_2O$ 

C. 
$$CaSO_4.5H_2O$$

D. 
$$CaSO_4H_2O$$

### **Answer: B**

**43.** Acetic acid reacts with NaOH to produce a salt, whose pH is

A. less than 7

B. equal to 7

C. more than 7

D. equal to 14

**Answer: B** 



**44.** The gas liberated when a metal is treated with acid or base

- A. oxygen
- B. nitrogen
- C. chlorine
- D. hydrogen

#### **Answer: D**



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**45.** High ductility is observed in

A. Na
B. Fe
C. Au
D. Mg
Answer: C
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<b>46.</b> Metal which is found in its native state?
A. Pt
B. Cu
C. Fe

D. Ag

#### **Answer: C**



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**47.** When brass is exposed to air for a long period of time it turns to green due to the

- A. formation of  $Cu(OH)_2$
- B. formation of  $CuSO_4$
- C. formation of  $CuCO_3$
- D. formation of  $Cu_2O$

### **Answer: C**



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- 48. Alloy used for welding purpose consists of.
  - A. lead and tin
  - B. lead and bismuth
  - C. lead and antimony
  - D. lead and copper

#### **Answer: A**



<b>49.</b> Liquid non-metal among the following
A. Br
B. Te
C. Hg
D. I





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**50.** Brass is the alloy of .

A. Cu and Sn

- B. NI and Cr
- C. Cu and Zn
- D. Pb and Sn

## **Answer: C**

