

## **CHEMISTRY**

## **BOOKS - OMEGA PUBLICATION**

## **AMINES**

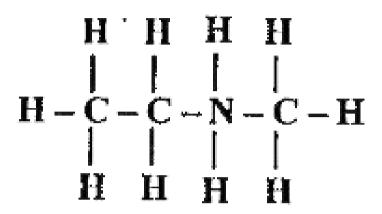
Questions

**1.** Give an example of primary amine and write its IUPAC name.



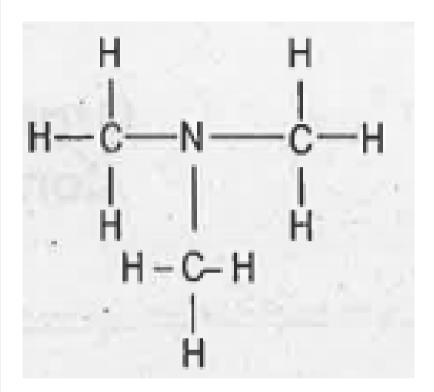
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2. Write the I.U.P.A.C name of 2° amine:





3. Write the IUPAC name of:





4. Write the structure of prop-2-en-1 amine.



**5.** How do cyanides and isocyanides differ in their structure ?



**6.** How will you convert nitrobenzene into aniline?



7. How is aniline prepared commercially?



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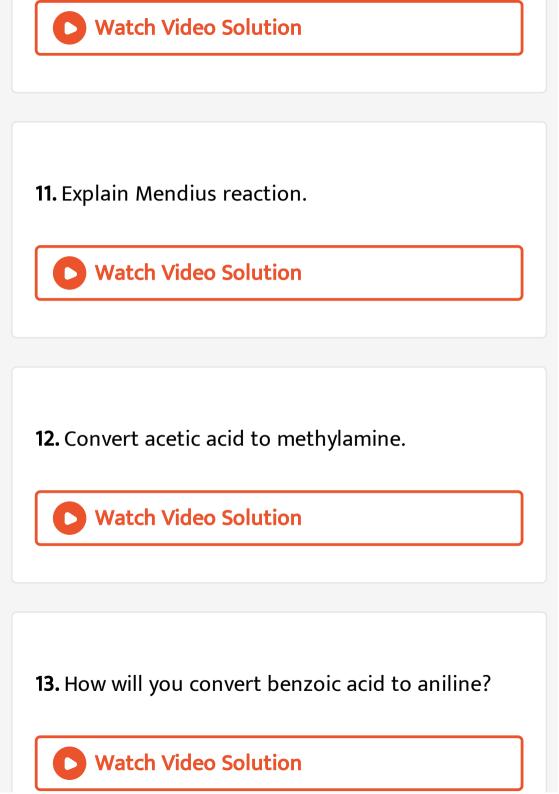
**8.** How will you convert nitrobenzene to benzene diazoninm chloride.



9. How will you convert benzene into aniline?



**10.** What happens when benzonitrile is treated with  $H_2$  gas and Ni ?



**14.** Write short note on Hoffmann's bromamide reaction. Why is it regarded as Hoffmann degradation reaction?



**15.** Convert the following : Propanoic acid to ethylamine.



**16.** Complete the following equations :

$$R-\stackrel{||}{C}-NH_2\stackrel{LiAlH_4}{\overset{LiAlH_4}{\longrightarrow}}$$



**17.** Complete the following equations :

$$C_6H_5NH_2+Br_2(aq)
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18. Write short note on ammonolysis.



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**19.** Pure amines cannot be prepared by ammonolysis. Why?



**20.** Mention the chief use of quaternary ammonium salts derived from long chain amines.



**21.** Gabriel phthalimide synthesis is used for the preparation of aromatic primary amines.



**22.** Why aromatic amines cannot be prepared by Gabriel phthalimide synthesis?



23. Convert nitrobenzene to acetanilide.



**24.** Why amines have lower boiling points than alcohols of comparable molecular masses?



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**25.** Why are the boiling points of carboxylic acids higher than the corresponding alcohols ?



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**26.** Arrange the following compounds in the increasing order of their boiling points :





## 27. Arrange the followng

Aniline, p-nitroaniline, p-methylaniline - in increasing of their basic strength.



**28.** Why amines have lower boiling points than alcohols of comparable molecular masses?



**29.** Explain why boiling point of ethylamine is less than that of ethyl alcohol .



**30.** Why do primary amines have higher boiling point than tertiary amines?



**31.** Why do primary amines have higher boiling point than tertiary amines?



**32.** Ethylamine is soluble in water whereas the aniline is not. Why?



33. Explain by giving reason:

Arylamines (aniline) are weaker bases than alkylamines (ethylamine) and ammonia.



**34.** Why ethylamine is more basic than methyl amine.



**35.** Give reasons fer the following:

o-nitrophenol is more acidic than o-methoxyphenol.



**36.** Account for the correct order oi' decreasing basicity of ethylamine, 2- amino ethanol, and 3-amino-1-propanol.



**37.** How  $1^{\circ}$  amine reacts with nitrous acid ?



**38.** How  $2^{\circ}$  amine reacts with nitrous acid ?



**39.** Why  $2^{\circ}$  amine is more basic than  $1^{\circ}$  and  $3^{\circ}$  amine ?



**40.** Glycine exists as  $H_3N^+CH_2COO^-$  , zwitter ion. Why?



**41.** Arrange the following in decreasing order of their basic strength in aqueous solution.

 $C_6H_5NH_2$ .  $C_2H_5NH_2$ ,  $(C_2H_5)_2NH$ ,  $NH_3$ 



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 $C_6H_5NH_2$ .  $C_2H_5NH_2$ ,  $(C_2H_5)_2NH$ ,  $NH_3$ 



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43. Why do amines act as nucleophiles?



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**44.**  $pK_b$  of methylamine is less than that of aniline.

Explain



**45.** Why ethylamine is more basic than aniline?



**46.** Explain, why amines are less acidic in nature than comparable alcohols ?



**47.** Explain, why amines are less acidic in nature than comparable alcohols ?



**48.** Aniline is a weaker base than cyclohexylamine. Why?



**49.** Which ismore basic, aliphatic amines or ammonia and why?



**50.** Why Methyl amine is stronger base than ammonia?



**51.** Why ethylamine is more basic than aniline?



**52.** In aqueous solution, trimethylamine is more basic than methylamine.

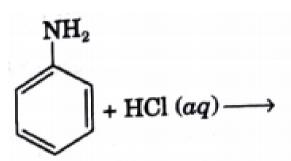


**53.** Write the reactions of aromatic primary amines with nitrous acid.



**54.** Write the reactions of aliphatic primary amines with nitrous acid.

**55.** Complete the following reaction :





**56.** 

What is Hinsberg's reagent?



**57.** Complete the following reaction:

$$C_6H_5N_2Cl \stackrel{CuCN}{\longrightarrow} A \stackrel{(\,H_2O\,)\,H^{\,+}}{\longrightarrow} B$$



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**58.** Give one chemical test to distinguish between the following pairs of compounds.

Methylamine and dimethylamine



**Watch Video Solution** 

**59.** Give one chemical test to distinguish between the following pairs of compounds.

Secondary and tertiary amines



**Watch Video Solution** 

**60.** Give one chemical test to distinguish between the following pairs of compounds.

Ethylamine and aniline



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**61.** Give a chemical test to distinguish between aniline and N-methyl aniline.



**62.** Name the test with which primary, secondary and tertiary amines can be distinguished.



**63.** What is Carbylamine reaction? How is it used to distinguish primary amines from secondary and

tertiary amines?



**64.** What happens when phenol is treated with benzoyl chloride in the presence of aqueous NaOH?



**65.**  $KMnO_4 \xrightarrow{\Delta} ? + MnO_2 + ?$ 



**66.** Which of the following amines undergo acylation reaction ?



67. Write short note on carbylamine reaction.



**68.** How can you find out whether a given amine is a primary amine? Write the chemical reaction involved in the test you perform.



**69.** What happens when ethylamine is warmed with chloroform and alcoholic !<,OH ?



**70.** How will you convert ethylamine into ethylisocyanide?



**71.** How will you convert Nitrobenzene to benzoic acid **Watch Video Solution 72.** How will you convert benzoic acid to aniline?

**Watch Video Solution** 

**73.** How will you convert

Benzamide to toluene



**74.** Accomplish the following conversions?

Aniline to benzyl alcohol.



**75.** How will you convert :

Aniline into p-bromoaniline



**76.** How  $3^{\circ}$  amine reacts with nitrous acid?



77. How are the following conversions accomplished? Write reaction only.

Aniline into chlorobenzene



**78.** How are the following conversions accomplished? Write reaction only.

Aniline into benzoic acid.

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**79.** Express the following in the scientific notation : 0.0048



**80.** Express the following in the scientific notation: 23405



**81.** Write the empirical formula of the following compound having the molecular formula C2H2.



82. How do you convert ethanol to propane nitrile.



**83.** Discuss the following reactions:

Nitration of aniline



**84.** Discuss the following reactions : Sulphonation of aniline



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**85.** Write the empirical formula of the compounds having the molecular formulas : N2O4



**86.** Write the empirical formula of the compounds having the molecular formulas : C6H6



**87.** Write the empirical formula of the compounds having the molecular formulas : H2O2



**88.** Write the empirical formula of the compounds having the molecular formulas : C12H22O11



**89.** Write the empirical formula of the compounds having the molecular formulas : C6H12O6



**90.** Write the empirical formula of the compounds having the molecular formulas : C10H8



**91.** What happens when ice cold solution of benzenediazonium chloride is treated with aniline?

Give chemical reaction also. **Watch Video Solution 92.** Write the empirical formula of the compounds having the molecular formulas: Na2CO3 **Watch Video Solution** 93. Write the empirical formula of the compounds having the molecular formulas: B2H6 **Watch Video Solution** 

**94.** Write the empirical formula of the compounds having the molecular formulas : N2O5



95. How will you convert aniline to benzene?



96. How will you convert

Methanamine to ethanamine.



**97.** How will you convert nitrobenzene into aniline ?



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98. How will you convert

Propionamide-to ethylamine.



**99.** Write the empirical formula of the compounds having the molecular formulas : Fe2O3



**Watch Video Solution** 

**100.** Write the iupac name of element with atomic number 128



**Watch Video Solution** 

**101.** How does benzene diazonium chloride react with:

Copper cyanide?



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**102.** How doesbenzene diazonium chloride react with:

 $Cu_{2}Br_{2}/HBr$ 



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103. Write the structures of main products when benzene diazonium chloride  $\left(C_6H_5N_2^+Cl^ight)$ 

reacts with the following reagents:

 $H_2O$ 



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**104.** Write the structures of main products when benzene diazonium chloride  $\left(C_6H_5N_2^+Cl^-\right)$  reacts with the following reagents :

 $CH_3CH_2OH$ 



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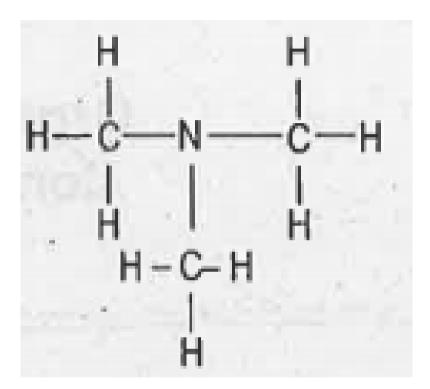


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$$R-\stackrel{||}{C}-NH_2\stackrel{LiAlH_4}{\longrightarrow}_{H_2O}$$



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$$C_6H_5NH_2+Br_2(aq)
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**123.** Pure amines cannot be prepared by ammonolysis. Why?



**124.** Mention the chief use of quaternary ammonium salts derived from long chain amines.



**125.** Give Gabriel phthalimide reaction for preparation of primary amine.



**126.** Why aromatic amines cannot be prepared by Gabriel phthalimide synthesis?



127. Convert nitrobenzene to acetanilide.



**128.** Why amines have lower boiling point than alcohols and carboxylic acids of comparable molecular masses ?



**129.** Why boiling point of alkyl cyanide are higher than those of isomeric isocyanides ?



130. Arrange the followng

 $C_2H_5NH_2,\,C_2H_5OH,\,(CH_3)_3N$ - in increasing order of their boiling point.



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Arylamines (aniline) are weaker bases than alkylamines (ethylamine) and ammonia.



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**141.** How  $1^{\circ}$  amine reacts with nitrous acid ?



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**143.** Explain why secondary amines are more basic than primary amines.



**144.** Glycine exists as  $H_3N^+CH_2COO^-$  , zwitter ion. Why?



**145.** Arrange the following compounds in an increasing order of basic strengths in their aqueous solutions :

 $NH_3, CH_3NH_2, (CH_3)_2NH, (CH_3)_3N$ 



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Watch Video Solution

147. Why do amines act as nucleophiles?



**148.**  $pK_b$  of aniline is more than that of metltylamine. Explain



**149.**  $pK_b$  of methylamine is less than that of aniline. Explain



**150.** Why are amines less acidic than alcohols?



**151.** Explain, why amines are less acidic in nature than comparable alcohols ?



**152.** Aniline is a weaker base than cyclohexylamine. Why?



**153.** Which is more basic, aliphatic amines or ammonia and why?



**154.** Give possible explanation that aliphatic amines are stronger bases than ammonia.



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**156.** Why ethylamine is more basic than methyl amine.



**157.** Write the reactions of aromatic primary amines with nitrous acid.



**158.** Write the reactions of aliphatic primary amines with nitrous acid.



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$$C_2H_5-NH_2+CHCl_3+3K-OH(alc)\stackrel{
m heat}{\longrightarrow}$$
 ?

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What is Hinsberg's reagent?



**161.** Complete the following reaction :

$$C_6H_5N_2Cl \stackrel{CuCN}{\longrightarrow} A \stackrel{H_2rac{\emptyset}{H^+}}{\longrightarrow} B$$



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**168.** How does aniline react with acetyl chloride in the presence of pyridine ?



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**169.** Fill in the blanks :  $C_6H_5NH_2 \xrightarrow[\Delta]{CH_3COCl}$  ?



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Watch Video Solution

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**Watch Video Solution** 

**181.** How are the following conversions accomplished? Write reaction only.

Aniline into chlorobenzene **Watch Video Solution 182.** How are the following conversions accomplished? Write reaction only. Aniline into benzoic acid.

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**183.** How aniline is converted to phenyl isothiocyanate?



**184.** How will you convert aniline into phenol?



**185.** How will you convert methyl bromide into ethane.



**186.** How do you convert ethanol to propane nitrile.



**187.** Discuss the following reactions:

Nitration of aniline



**188.** Discuss the following reactions:

Sulphonation of aniline



**189.** Aniline does not undergo Friedel-Crafts reaction. Explain.



190. What is T.N.T.? Give its structure.



191. Write short note on diazotisation reaction.



192. Write short note on Sandmeyer reactions.

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**193.** Explain the following reaction:

Gattermann reaction.



194. Write short notes on Coupling reaction



195. What happens when ice cold solution of benzenediazonium chloride is treated with aniline? Give chemical reaction also.



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**196.** Complete the reaction

$$C_6H_5\overset{+}{N_2}\overset{-}{C}l + H_3PO_2 + H_2O 
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**197.** How will you convert Nitrobenzene to benzoic acid **Watch Video Solution 198.** How will you convert Benzamide to toluene **Watch Video Solution** 

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**201.** How will you convert nitrobenzene into aniline

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**203.** Write the reaction of benzene diazonium chloride with :

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204. Write the reaction of benzenediazonium chloride with



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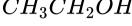
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**207.** Write the structures of main products when benzene diazonium chloride  $\left(C_6H_5N_2^+Cl^ight)$  reacts with the following reagents :

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208. Write the structures of main products when benzene diazonium chloride  $\left(C_6H_5N_2^+Cl^ight)$ reacts with the following reagents:  $CH_3CH_2OH$ 





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# **Multiple Choice Questions**

1. Stephen reduction converts cyanides to

A. Aldehydes

**B.** Ketones

- C. Amines
- D. Acids

# **Answer: A**



- **2.** Classify the following as pure substances or mixtures:
  - A. a. graphite
  - B. b. milk
  - C. c. air

D. d. diamond

**Answer: B** 



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**3.** Identify the following as homogenous and hetrogenous mixture:

A. sugar dissolved in water

B. oil and water

C.

D.

# **Answer: A**



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**4.** Which of the following is not an ambident nucleophile?

A. 
$$NO_2^-$$

B. 
$$OH^-$$

C. 
$$SCN^-$$

D. 
$$CN^-$$

Answer: B

**5.** Which of the following substances on treatment with  $P_2 O_5$  gives ethanenitrile ?

A. Propanamide

B. Ethanamide

C. Ethanoic acid

D. N-methylethylamine

**Answer: B** 



**6.** Methyl cyanide on treatment with methyl magnesium bromide followed by hydrolysis gives

A. propanone

B. ethanol

C. ethanal

D. propanal

**Answer: A** 



**7.** Which of the following represents the poisonous gas which caused tragedy in Bhopal in 1984?

A. 
$$CH_3C=N=S$$

B. 
$$CH_3 \equiv C = O$$

$$\operatorname{D.} CH_3 - O - N = C$$

### **Answer: B**



**8.** When propane is subjected fo the treatment with fuming nitric acid at 673 K which of the following will not form?

A. 1-nitropropane

B. 1-nitropropane

C. Nitromethane

D. Nitrohexane

### **Answer: D**



9. Which of the following is not a nitro derivative?

A.  $C_6H_5NO_2$ 

B.  $CH_3CH_2ONO$ 

C. CH,CH-N

D.  $C_6H_4(OH)NO_2$ 

## **Answer: B**



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**10.** Which of the following structures represent nitrolic acid?

A. 
$$R_2C=NOH$$

$$\text{B.}\,R_2CH - CH - COOH \\ | \\ CN$$

C. 
$$R_2 \stackrel{C}{\underset{NO_2}{\mid}} = NOH$$

D. 
$$R_2N-N=O$$

### **Answer: C**



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**11.** Which reagent should be employed to get ethyl carbylamine from ethyl iodide as major product?

A. HCN

B. AgCN

C. KCN

D.  $AgNO_2$  followed by reduction.

# Answer: B



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**12.**  $NO_2^+$  is called

A. nitronium ion

B. nitrosonium ion

C. nitro group

D. none of above

# **Answer: A**



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# **13.** Explain Mendius reaction.

A. RCN and  $Na/C_2H_5OH$ 

B. RCN and  $H_2O/HCl$ 

C. RCN and  $Zn/NH_4Cl$ 

D. RCN and  $SnCl_2 \, / \, HCl$ 

## **Answer: A**



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**14.** Aniline upon heating with conc. $HNO_3$  and conc. $H_2SO_4$ 

mixture gives:

- A. o-and p-nitroaniline
- B. o-nitroaniline
- C. black tarry mass
- D. No reaction

**Answer: A** 



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15. Which are given by carbylamine reaction?

A.  $CH_3NH_2$ 

B.  $C_2H_5NO_2$ 

C.  $CH_3CONH_2$ 

D.  $(CH_3)_2NH$ 

**Answer: A** 



**16.** Nitrobenzene is subjected to reductfon with zinc dust and ammonium chloride. The main product formed will be

A. benzenamine

B. aniline

C. N-phenylhydroxylamine

D. none of above

**Answer: C** 



17. A primary nitroalkane is treated with nitrous acid, which of the following will be the main product.

- A. Pseudonitrole
- B. Nitrolic acid
- C. A primary amine
- D. Primary alcohol

### **Answer: B**



**18.** Nitromethane is subjected to the treatment with chlorine in the presence of sodium hydroxide, the main product is

A. monochloronitromethane

B. trichloromethane

C. chloropicrin

D. none of above.

**Answer: C** 



**19.** The electrolytic reduction of nitrobenzene in strongly acidic medium produces

A. Phenol

B. p-Aminophenol

C. Hydroazobenzene

D. Azobenzene

Answer: B



**20.** Which of the following on boiling with  $Na_2CO_3$ (aq) gives aniline?

A. Nitrobenzene

B. Anilinium chloride

C. Chlorobenzene

D. Benzene diazonium chloride.

**Answer: B** 



**21.** Which of the following reagents on teatment with benzenamine in basic medium produces phenyl isocyanide?

- A.  $CCl_4$
- B. trichloromethane
- C. methylen dichloride
- D. hexachloroethane

### **Answer: B**



**22.** An organic compound with formula  $C_3H_5N$  on hydrolysis forms an acid which reduces Fehling solution, The compound can be

- A. Ethanenitrile
- B. Isocyanoethane
- C. Ethoxyethane
- D. Propanenitrile

**Answer: B** 



23. The type of isomerism exhibited by

$$C_6H_5C\equiv N$$
 and  $C_6H_5N\equiv C$  is

- A. position
- B. functional
- C. enantiomerism
- D. functional as well as tautomerism

### **Answer: B**



24. In the following reaction,

$$C_6H_5NH_2+CHCl_3+3NaOH
ightarrow A+3B+3C$$

The product A is

A. phenyl isocyanide

B. phenyl isocyanide

C. ethyl chloride

D. HCl or  $H_2O$ 

#### **Answer: A**



25. When benzene diazonium chloride solution is
boiled, it yields
A. Benzene
B. Aniline
b. Amme
C. Phenol
D. Chlorobenzene





**26.** Stephen reduction converts cyanides to

- A. aldehydes
- B. ketones
- C. amines
- D. acids

# Answer: A



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**27.** Which of the following substances on treatment with  $P_2O_5$  gives ethanenitrile ?

A. Propanamide

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**28.** Aniline upon heating with conc. $HNO_3$  and  ${
m conc.}H_2SO_4$ 

mixture gives:

A. o-and p-nitroaniline

B. o-nitroaniline

C. black tarry mass

D. no reaction.

# **Answer: A**



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29. Which are given by carbylamine reaction?

A.  $CH_3NH_2$ 

B.  $C_2H_5NO_2$ 

C.  $CH_3CONH_2$ 

D.  $(CH_3)NH$ 

**Answer: A** 



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**30.** The electrolytic reduction of nitrobenzene in strongly acidic medium produces

A. phenol

B. p-Aminophenol

C. hydroazobenzene

D. azobenzene



**Watch Video Solution** 

**31.** Which of the following on boiling with  $Na_2CO_3$  (aq) gives aniline?

- A. Nitrobenzene
- B. Anilinium chloride
- C. Chlorobenzene
- D. Benzene diazonium chloride.

Answer: B

**32.** An organic compound with formula  $C_3H_5N$  on hydrolysis forms an acid which reduces Fehling solution, The compound can be

A. ethanenitrile

B. isocyanoethane

C. ethoxyethane

D. propanenitrile

**Answer: B** 



33. In the following reaction,

$$C_6H_5NH_2+CHCl_3+3NaOH
ightarrow A+3B+3C$$

The product A is

A. phenyl isocyanide

B. phenyl cyanide

C. ethyl chloride

D. HCl or  $H_2O$ 

#### **Answer: A**



**34.** Which reactants are involved in Mendius reaction?

A. RCN and  $Na/C_2H_5OH$ 

B. RCN and  $H_2O/\!/HCl$ 

C. RCN and  $Zn/NH_4Cl$ 

D. RCN and  $SnCl_2 \, / \, HCl$ 

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**35.** Nitrobenzene is subjected to reductfon with zinc dust and ammonium chloride. The main product formed will be

A. benzenamine

B. aniline

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D. none of above.

# **Answer: C**



**36.** Which compound undergoes Hoffmann 's

**Bromamide Reaction?** 

A. HCHO

B.  $CH_3CHO$ 

C.  $CH_3OH$ 

D. 
$$CH_3-\stackrel{|}{C}-NH_2$$

#### **Answer: D**



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**37.** Which one of the following is more basic?

A. 
$$C_6H_5-NH_2$$

B.  $NH_3$ 

 $\mathsf{C.}\,CH_3CH_2-NH_2$ 

D.  $CH3 - NH_2$ 

# **Answer: C**



**38.** Express the following numbers to four significant figures: 1.81234 (10\*3)



<b>39.</b> Stephen	reduction.	converts c	yanides to
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- A. Aldehydes
- **B.** Ketones
- C. Amines
- D. Acids

**Answer: A** 



**40.** The electrophile involved in the nitration of benzene is

- A.  $NO_2$
- $\mathsf{B.}\,NO_2^{\,+}$
- C. *NO*
- D.  $NO_2^-$

**Answer: B** 



- **41.** Among following statements on the nitration of aromatic compounds, the false one is
  - A. The fate of nitration of benzene is almost the same as that of hexadeuterobenzene.
  - B. The rate of nitration of toluene is greater than that of benzene.
  - C. The rate of nitration of benzene is greater than that of hexadeuterobenzene
  - D. Nitration is an electrophilic substitution reaction.

# Answer: A



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**42.** Which of the following is not an ambident nucleophile?

A. 
$$NO_2^-$$

B. 
$$OH^-$$

$$\mathsf{C.}\,SCN^{\,-}$$

D. 
$$CN^-$$

Answer: B

**43.** Which of the following substances on treatment with  $P_2O_5$  gives ethanenitrile ?

A. Propanamide

B. Ethanamide

C. Ethanoic acid

D. N-methylethylamine

**Answer: B** 



**44.** Methyl cyanide on treatment with methyl magnesium bromide followed by hydrolysis gives

- A. propanone
- B. ethanol
- C. ethanal
- D. propanal

**Answer: A** 



**45.** Which of the following represents the poisonous gas which caused tragedy in Bhopal in 1984?

A. 
$$CH_3C = N = S$$

B. 
$$CH_3 \equiv C = O$$

$$\mathsf{C.}\,CH_3-N=C=S$$

$$D. CH_3 - O - N = C$$

#### **Answer: B**



**46.** When propane is subjected fo the treatment with fuming nitric acid at 673 K which of the following will not form?

- A. 1-nitropropane
- B. 1-nitropropane
- C. Nitromethane
- D. Nitrohexane

**Answer: D** 



**47.** Which of the following is not a nitro derivative?

A.  $C_6H_5NO_2$ 

B.  $CH_3CH_2ONO$ 

CH,CH-N

D.  $C_6H_4(OH)NO_2$ 

# **Answer: B**



**48.** Which of the following structures represent nitrolic acid?

A. 
$$R_2C=\mathop{NOH}\limits_{\mid CN}$$

B. 
$$R_2CH - CH - COOH$$

C. 
$$R_2 \stackrel{C}{\underset{NO_2}{\mid}} = NOH$$

$$\mathsf{D}.\,R_2N-N=O$$

#### **Answer: C**



**49.** Which reagent should be employed to get ethyl carbylamine from ethyl iodide as major product?

A. HCN

B. AgCN

C. KCN

D.  $AgNO_2$  followed by reduction.

# **Answer: B**



**50.**  $NO_2^+$  is called

A. nitronium ion

B. nitrosonium ion

C. nitro group

D. none of above

# **Answer: A**



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**51.** Which reactants are involved in Mendius reaction?

A. RCN and  $Na \, / \, C_2 H_5 OH$ 

B. RCN and  $H_2O/HCl$ 

C. RCN and  $Zn\,/\,NH_4Cl$ 

D. RCN and  $SnCl_2 \, / \, HCl$ 

#### **Answer: A**



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**52.** Aniline upon heating with conc. $HNO_3$  and conc. $H_2SO_4$ 

mixture gives:

A. o-and p-nitroaniline

B. o-nitroaniline

C. black tarry mass

D. No reaction

# Answer: A



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**53.** Which are given by carbylamine reaction?

A.  $CH_3NH_2$ 

B.  $C_2H_5NO_2$ 

C.  $CH_3CONH_2$ 

D.  $(CH_3)_2NH$ 

**Answer: A** 



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**54.** Nitrobenzene is subjected to reductfon with zinc dust and ammonium chloride. The main product formed will be

A. benzenamine

B. aniline

- C. N-phenylhydroxylamine
- D. none of above

**Answer: C** 



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**55.** A primary nitroalkane is treated with nitrous acid, which of the following will be the main product.

- A. Pseudonitrole
- B. Nitrolic acid

- C. A primary amine
- D. Primary alcohol



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**56.** Nitromethane is subjected to the treatment with chlorine in the presence of sodium hydroxide, the main product is

- A. monochloronitromethane
- B. trichloromethane

- C. chloropicrin
- D. none of above.

# **Answer: C**



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**57.** The electrolytic reduction of nitrobenzene in strongly acidic medium produces

- A. Phenol
- B. p-Aminophenol
- C. Hydroazobenzene

D. Azobenzene

**Answer: B** 



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**58.** Which of the following on boiling with  $Na_2CO_3(\mathsf{aq})$  gives aniline?

- A. Nitrobenzene
- B. Anilinium chloride
- C. Chlorobenzene
- D. Benzene diazonium chloride.



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**59.** Which of the following reagents on teatment with benzenamine in basic medium produces phenyl isocyanide?

- A.  $CCl_4$
- B. trichloromethane
- C. methylen dichloride
- D. hexachloroethane



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**60.** An organic compound with formula  $C_3H_5N$  on hydrolysis forms an acid which reduces Fehling solution, The compound can be

- A. Ethanenitrile
- B. Isocyanoethane
- C. Ethoxyethane
- D. Propanenitrile



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61. The type of isomerism exhibited by

$$C_6H_5C\equiv N$$
 and  $C_6H_5N\equiv C$  is

A. position

B. functional

C. enantiomerism

D. functional as well as tautomerism

**Answer: B** 

**62.** In the following reaction,

 $C_6H_5NH_2+CHCl_3+3NaOH
ightarrow A+3B+3C$ 

The product A is

A. phenyl isocyanide

B. phenyl isocyanide

C. ethyl chloride

D. HCl or  $H_2 O$ 

# Answer: A

**63.** When benzene diazonium chloride solution is boiled, it yields

A. Benzene

B. Aniline

C. Phenol

D. Chlorobenzene

**Answer: C** 



**64.** Stephen reduction converts cyanides to

A. aldehydes

B. ketones

C. amines

D. acids

# Answer: A



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**65.** Which of the following substances on treatment with  $P_2 O_5$  gives ethanenitrile ?

- A. Propanamide
- B. Ethanamide
- C. Ethanoic acid
- D. N-methylethylamine.

### **Answer: B**



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**66.** Aniline upon heating with conc. $HNO_3$  and conc. $H_2SO_4$ 

mixture gives:

A. o-and p-nitroaniline

B. o-nitroaniline

C. black tarry mass

D. no reaction.

# **Answer: A**



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67. Which are given by carbylamine reaction?

A.  $CH_3NH_2$ 

B.  $C_2H_5NO_2$ 

 $\mathsf{C.}\,\mathit{CH}_3\mathit{CONH}_2$ 

D.  $(CH_3)NH$ 

**Answer: A** 



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**68.** The electrolytic reduction of nitrobenzene in strongly acidic medium produces

A. phenol

B. p-Aminophenol

C. hydroazobenzene

D. azobenzene

### **Answer: B**



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**69.** Which of the following on boiling with  $Na_2CO_3$  (aq) gives aniline?

A. Nitrobenzene

B. Anilinium chloride

C. Chlorobenzene

D. Benzene diazonium chloride.

### **Answer: B**



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**70.** An organic compound with formula  $C_3H_5N$  on hydrolysis forms an acid which reduces Fehling solution, The compound can be

- A. ethanenitrile
- B. isocyanoethane
- C. ethoxyethane
- D. propanenitrile

# **Answer: B**



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71. In the following reaction,

$$C_6H_5NH_2+CHCl_3+3NaOH
ightarrow A+3B+3C$$

The product A is

A. phenyl isocyanide

B. phenyl cyanide

C. ethyl chloride

D. HCl or  $H_2{\cal O}$ 

# **Answer: A**



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**72.** Which reactants are involved in Mendius reaction?

A. RCN and  $Na/C_2H_5OH$ 

B. RCN and  $H_2O/\!/HCl$ 

C. RCN and  $Zn\,/\,NH_4Cl$ 

D. RCN and  $SnCl_2 \, / \, HCl$ 

**Answer: A** 

73. Nitrobenzene is subjected to reductfon with zinc dust and ammonium chloride. The main product formed will be

A. benzenamine

B. aniline

C. N-phenylhydroxylamine

D. none of above.

**Answer: C** 



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**74.** Which compound undergoes Hoffmann 's Bromamide Reaction?

B. 
$$CH_3CHO$$

$$\mathsf{C}.\,CH_3OH$$

D. 
$$CH_3 - C - NH_2$$

#### **Answer: D**



75. Which one of the following is more basic?

A. 
$$C_6H_5-NH_2$$

B.  $NH_3$ 

C. 
$$CH_3CH_2 - NH_2$$

D. 
$$CH3-NH_2$$

## **Answer: C**



**76.** Benzene diazonium chloride on reaction with phenol in basic medium gives:

- A. Diphenyl ether
- B. p-Hydroxy azobenzene
- C. Chlorobenzene
- D. Benzene

**Answer: D** 

