

PHYSICS

BOOKS - BETTER CHOICE PUBLICATION

RADIOACTIVITY

Very Short Answer Type Questions

1. What is the difference between an electron and a β -particle?



2. Define decay constant.



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3. Define decay constant.



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4. What are α -particles?



5. What is β -particle?



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6. Define beta (β) decay.



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7. Define gamma (γ) decay.



8. Define alpha (α) decay



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9. Among α (alpha), β (beta) and γ (gamma) rays which one has highest penetrating power



10. Among α (alpha), β (beta) and γ (gamma) rays which one has highest penetrating power ?



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11. Define Radioactivity of a radioactive substance.



12. Which has greater ionising power : alpha particle or beta particle ?



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13. Among α (alpha), β (beta) and γ (gamma) radiations, which one is not affected by a magnetic field ?



1. The neutron to proton ratio increases during β -decay. (True /false)



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2. Define half life of a radioactive substance.





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4. State the law of radioactive decay. Show that radioactive decay is exponential in nature.





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Long Answer Type Questions

1. State the law of radioactive decay. Show that radioactive decay is exponential in nature.





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3. State the law of radioactive decay. Show that radioactive decay is exponential in nature.





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Numericals Problems

1. The half life of a radioactive element X is 500 years. In how many years its activity will decrease by 90% of its initial value.



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2. The half life of a radioactive substance is 200 years. In how many years, its activity will decay to 0.1 times of its initial value.



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3. The half life of a radioactive element is 400 years. In how many years, its activity will decay to $\frac{1}{10}$ th of its initial value ?



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4. Half life of a certain radioactive material is 130 days. After what lapse of time, the undecayed fraction of the material will be 25%?



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5. Half life of a certain radioactive material is 140 days. After what lapse of time, the undecayed fraction of the material will be 25%?



Most Expected Questions

1. What are γ -rays?



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2. Why do alpha particles have a hight ionising power?



3. What are radio-isotopes? What are their applications?



4. What do you mean by activity of a radioactive substance?



5. Natural radioactive nuclei are nuclei of high mass number. Why?



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6. A nucleus contains no electrons, but can eject them. Why?

