

CHEMISTRY

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CHEMISTRY IN EVERYDAY LIFE

Evaluation Textbook Questions Answers Choose
The Correct Answer

1. Which of the following is an analgesic?

- A. Streptomycin
- B. Chloromycetin
- C. Aspirin
- D. Penicillin

Answer: C



- 2. Dettol is the mixture of.
 - A. chloroxylenol and bithionol

- B. chloroxylenol and a-terpineol
- C. phenol and iodine
- D. terpineol and bithionol

Answer: B



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3. Antiseptics and disinfectants either kill or prevent growth of microorganisms. Identify which of the following statement is not true.

A. Dilute solutions of boric acid and hydrogen peroxide are strong antiseptics.

B. Disinfectants harm the living tissues.

C. A0.2% solution of phenol is an antiseptic while 1% solution acts as a disinfectant.

D. Chlorine and iodine are used as strong disinfectants

Answer: A



4. Saccharin, an artificial sweetener is manufactured from:

A. cellulose

B. toluene

C. cyclohexene

D. starch

Answer: B



5. Drugs that bind to the receptor site and inhibit its natural function are called:

A. antagonists

B. agonists

C. enzymes

D. molecular targets

Answer: A



6. Aspirin is a/an:

A. acetylsalicylic acid

B. benzoyl salicylic acid

C. chlorobenzoic acid

D. anthranilic acid

Answer: A



7. Which one of the following structures represents nylon 6,6 polymer?





Answer: D



8. Natural rubber has:

A. alternate cis- and trans-configuration

B. random cis- and trans-configuration

C. all cis-configuration

D. all trans-configuration

Answer: C



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9. Nylon is an example of

- A. polyamide
- B. polythene
- C. polyester
- D. poly saccharide

Answer: A



- **10.** Terylene is an example of.
 - A. polyamide

B. polythene

C. polyester

D. polysaccharide

Answer: C



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11. Which is the monomer of neoprene in the following?

A.
$$CH_2 - CH - CH = CH_2$$

$$\mathsf{B.}\,CH_2=CH-C\equiv CH$$

$$\mathsf{C.}\,CH_2=CH-CH=CH_2$$

D.
$$CH_2 = C - CH = CH_2$$
 $_{Cl_3}$

Answer: A



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12. Which one of the following is a bio - degradable polymer?

A. HDPE

- B. PVC
- C. Nylon 6
- D. PHBV

Answer: D



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13. Non stick cook wares generally have a coating (a) antiseptic (b) antipyretic of a polymer, whose monomer is:

- A. ethane
- B. prop-2- enenitrile
- C. choroethene
- D. 1,1,2,2-tetrafluoroethane

Answer: D



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14. Assertion: 2-methyl-1,3-butadiene is the monomer of natural rubber.

Reason: Natural rubber is formed through anionic addition polymerisation.

A. If both assertion and reason are true and reason is the correct explanation of assertion.

B. If both assertion and reason are true but reason is not the correct explanation of assertion

C. Assertion is true but reason is false.

D. Both assertion and reason are false.

Answer: C



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15. An example of antifertility drug is:

A. novestrol

B. seldane

C. salvarsan

D. chloramphenicol

Answer: A

16. The drug used to induce sleep is:

A. paracetamol

B. bithional

C. chloroquine

D. equanil

Answer: D



17. Which of the following is a co-polymer?

A. Orlon

B. PVC

C. Teflon

D. PHBV

Answer: D



18. The polymer used in making blankets (artificial wool) is:

A. polystyrene

B. PAN

C. polyester

D. polythene

Answer: B



- **19.** Regarding cross-linked or network polymers, which of the following statement is incorrect?
 - A. Examples are Bakelite and melamine.
 - B. They are formed from bi and trifunctional monomers.
 - C. They contain covalent bonds between various linear polymer chains.
 - D. They contain strong covalent bonds in their polymer chain.

Answer: D



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20. Amixture of chloroxylenol and terpinecol acts as:

- A. antiseptic
- B. antipyretic
- C. antibiotic
- D. analgesic

Answer: A



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Evaluation Textbook Questions Answers Answer The Following Questions

1. Which chemical is responsible for the antiseptic properties of dettol?



2. What are antibiotics?

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3. Name one substance which can act as both analgesic and antiphyretic.



4. Write a note on synthetic detergents.



5. How do antiseptics differ from disinfectants?



6. What are food preservatives?



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7. Why do soaps not work in hard water?



8. What are drugs? How are they classified?



9. How the tranquilizers work in body?



10. Write the structural formula of aspirin.

11. Explain the mechanism of cleansing action of soaps and detergents.



12. Which sweetening agent are used to prepare sweets for a diabetic patient?



13. What are narcotic and non-narcotic drugs.

Give examples



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14. What are antifertility drugs? Give examples.



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15. Write a note on co-polymer.



16. What are bio degradable polymers? Give examples.



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17. How is terylene prepared?



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18. Write a note on vulcanization of rubber.



19. Classify the following as linear, branched or cross linked polymers

(i) Bakelite (ii) Nylon (iii) Polythene



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20. Differentiate thermoplastic and thermosetting polymers



21.	Which	one	of	the	following	antacids	is	an

(b) tranquilizers antihistamine?

A. Ranitidine

B. Lansoprazole

C. Terfen adine

D. Luminal

Answer: A



22. Which of the following is / are nurologically active drug?

A. Aspirin

B. Phenelzine

C. Heroin

D. all the above

Answer: D



23. Antiseptic chloroxylenol is:

- A. 4- chloro, 3, 5 dimethyl phenol
- B. 3 chloro, 4, 5 dimethyl phenol
- C. 4-chloro, 2, 5 dimethyl phenol
- D. 5- chloro, 3, 4 dimethyl phenol

Answer: A



24. Structurally a biodegradable detergent should Ans: (c) contain a:

A. normal alkyl chain

B. phenyl side chain

C. cyclohexyl side chain

D.

Answer: A



25. Which of the following statements is not correct?

- A. Some antiseptics can be added to soap.
- B. Dilute solutions of disinfectants can be used as antiseptic.
- C. Disinfectants are antimicrobial drugs.
- D. Antiseptic medicine can be infected.

Answer: D



Other Important Questions Answers Choose The Correct Answer

1. The most useful classification of drugs for medicinal chemists is:

A. on the basis of chemical structure

B. on the basis of drug action

C. on the basis of molecular targets

D. on the basis of active drug.

Answer: C

2. A compound that causes general antidepressant action on the central nervous system belongs to the class of

A. analgesics

B. tranquilizers

C. narcotic analgesics

D. antihistamines

Answer: B

3. Compound which is added to soap to inpart antiseptic properties is:

A. sodium lauryl sulphate

B. sodium do decyl benzene sulphonate

C. resin

D. bithional

Answer: D



4. Glycerol is added to soap. Its function is:

A. as a filler

B. to increase lathering

C. to prevent rapid drying

D. to make soap granules

Answer: C



5. Polyethylene glycols are used in the preparation of which type of detergents?

A. Cationic detergents

B. Anionic detergents

C. Non-ionic detergents

D. Soaps

Answer: C



6. Which of the following is employed as anti-histamine?

A. Omeprazole

B. Chloroampinicol

C. Diphenylhydramine

D. Norethindrone

Answer: C



7. Tincture of iodine is:

A. aqueous solution of I_2

B. solution of I_2 in KI

C. alcoholic solution of I_2

D. aqueous solution of KI

Answer: C



8. Which among the following is not an antibiotic?

A. Penicillin

B. Oxytocin

C. Erythromycin

D. Tetracyclin

Answer: B



9. Which of the following is not an antimicrobial?

A. Salvarsan

B. Sulphanilamide

C. Prontsil

D. Paracetamol

Answer: D



10. Which of the following is not a semi synthetic polymer?

- A. Cis poly isoprene
- B. Cellulose nitrate
- C. Cellulose acetate
- D. Vulcanised rubber

Answer: A



11. Which of the following polymers is prepared by condensation polymerisation?

- A. Styrene
- B. Nylon 6,6
- C. Teflon
- D. Rubber

Answer: B



12. Which of the following is a chain growth polymer?

A. Starch

B. Nucleic acid

C. Polystyrene

D. Proteins

Answer: C



13. Terelene is a condensation polymer of ethylene glycol and:

A. benzoic acid

B. pthalic acid

C. salicylic acid

D. terepthalic acid

Answer: D



14. Which one of the following is a copolymer formed by condensation polymerisation?

- A. Tereleme
- B. Buna-S
- C. Buna- N
- D. Neoprene

Answer: A



15. Bakelite is obtained from phenol by the reaction with:

A. HCHO

B. $(CH_2OH)_2$

C. CH_3CHO

D. CH_3COCH

Answer: A



16. Which of the following statements is not true?

A. Buna - S is a copolymer of butadiene and styrene.

B. Natural rubber is a1,4-polymer of isoprene.

C. In vulcanisation the formation of sulphur bridges between different

chains makes rubber harder and stronger.

D. Natural rubber has trans configuration at every double bond.

Answer: D



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17. Teflon, styron and neoprene are all:

A. copolymers

- B. condensation polymers
- C. homo polymers
- D. monomers

Answer: C



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18. Which of the following sets contain only thermoplastics?

A. Polythene, bakelite, nylon-6

- B. Glyptal, melane, PAN
- C. PVC, PMMA, Polystyrene
- D. Polypropylene urea formaldehyde, teflon

Answer: C



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19. Which of the following sets contain only copolymers?

A. SBR, Glyptal, Nylon 6,6

- B. Nylon 6, Butyl rubber, Neoprene
- C. Poly ethylene, polyester, PVC
- D. Melmac, Bakelite, Teflon.

Answer: A



- **20.** Which of the following are not
 - thermosetting polymers?
- (1) Bakelite (2) Polystyrene
- (3) PVC (4) Melmac

- A. 1, 2
- B. 2, 3
- C. 2, 4
- D. 3, 4

Answer: B



21. Define the term chemotherapy



22. Define therapeutic index? What is its use?



23. Explain the term, target molecules or drug targets as used in medicinal chemistry.



24. Give examples of drugs which are grou ed based on the biological effect that they

produce on the recipient.



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25. Streptomycin and erythromycin are classified in the same group. Justify the statement.



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Other Important Questions Answers Answer The Following Questions

1. Write a short note on enzymes as drug targets.



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2. How do drugs interact with targets? (or) Give a brief account of drug target interaction.



3. How do drugs interact with enzymes? (or) Give a brief account of drug - enzyme interaction.



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4. Explain the terms, (i) Competitive inhibitors, (ii) Allosteric inhibitors.



5. Explain the term, (i) Chemical messengers (ii) Receptor, (iii) Antagonists, (iv) Agonists.



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6. Mention the various types of chemical messengers and explain how they act ?



7. With reference to which classification has the statement "ranitidine, is an antacid" been given?



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8. List two major classes of antibiotics with an example of each class.



9. What are tranquilizers? How do they act? Give two examples.



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10. Give examples each (i) Anti inflammatory drugs (ii) Antipyretics (iii) non steroidal Anti inflammatory drugs.



11. Mention an important difference between a non-narcotic analgesics and a narcotic analgesics.



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12. Give two examples for narcotic analgesics.



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13. Give two examples for local anesthetics.

14. Name the anesthetics used for major surgical procedures.



15. Give examples of antacids. How do antacids function in case of acidity?



16. What are antihistamines? Give two examples.



17. What are antimicrobials? Give two examples



18. How does (i) β lactams and (ii) macro-lides function as antimicrobials.



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19. Give examples for β - lactam and macrolides antimicrobials.



20. What are the uses of β -lactam and macrolide antimicrobials.



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21. Give examples for 'Fluoroquinolones' and mention their uses.



22. How does tetracyclines class of antibiotics function'? Mention their uses.



23. What are aminoglycosides? Give examples.



24. What are food additives? Give examples.



25. What are food preservatives?



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26. Name the preservative used in the preparation of pickles and vegetables.



27. Name the chemicals which are used as emulsifiers.



28. Name the physical methods used in the preservative of food.



29. What are antioxidants? Give examples.



30. How do antioxidants prevent the oxidation of food?



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31. Define "Total Fatty Matter (TFM). What is its use?



32. What are anionic detergent? **View Text Solution** 33. What are cationic detergent? **View Text Solution 34.** What are Non-ionic detergent?

35. Explain the term monomer and polymer and polymerisation.



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36. What are synthetic and natural polymers? Give two examples for each type.



37. In which classes, the polymers are classified on the basis of molecular forces.



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38. Write names of monomers of the following polymers and classify then as addition or condensation polymers.

(a) Teflon, (b) Bakelite, (c) Natural rubber



39. What is the role of benzoyl peroxide in the polymerisation of ethene?



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40. What are LDPE and HDPE? How are they prepared?



41. Write the structure of the monomers of the Polymer Teflon following polymers. (i) PVC, (ii) Polypropene, (iii) PAN, (iv) Nylon - 6.



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42. Give examples each of (i) addition polymers (ii) condensation polymer (iii) copolymer.



43. Write the names of structure of monomers of the following polymers. (i) Buna - S, (i) Neoprene



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44. What is repeating unit in the condensation polymer by combining $HOOC-CH_2-CH_2-COOH \quad \text{(succinic acid) and HNCH,CH,NH (ethylene diamine)}.$



45. Differentiate between molecular structure and behaviour of thermoplastic and thermosetting plastic. Give one example each type.



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46. What are the monomeric repeating units of Nylon 6, Nylon 6, 6?



47. Name a synthetic polymer which is an amide.



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48. Mention which of the following are addition polymers. (i) Terelene, (ii) Nylon 6, 6 (iii) Neoprene, (iv) Teflon.



49. What are biodegradable polymers?



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50. How are polymer classification based on forces operating between their molecule. Classification based as sources:



51. Give the preparation of bakelite and its uses. Preparation of bakelite:



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52. Give the preparation and use of melamine.

Preparation of melamine:



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53. Give examples for biodegradable polymer.



54. How is decron obtained from ethylene glycol and trtrphthalic acid?



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55. How is urea formaldehyde prepared?



56. How is PHBV prepared? Give equation mention its uses.



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57. Identify the type of polymer where A and B are monomers.-A-B-B-A-A-B-A.



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58. Why is bakelite a thermosetting polymer?

59. Give a brief account of vulcanisation of rubber.



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60. Describe the preparation of neoprene and mention (styrene) its uses.



61. What is the name of the polymer formed from the monomers acrylonitrile and butal, 3-diene? How it is prepared?



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62. What type of polymer Buna-S is? Give its method of preparation

Preparation of Buna-S:



63. Name the polymer formed by the copolymerisation of glycine and amino caproic acid. How it is prepared?

