



CHEMISTRY

BOOKS - CBSE COMPLEMENTARY MATERIAL CHEMISTRY (HINGLISH)

ALDEHYDES, KETONES AND CARBOXYLIC ACIDS

Multiple Choice Questions



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is treated
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with concentrated KOH solution ?



Answer: B

2.

$$CH_3-CH_2-C\equiv CH \stackrel{40\,\%\,H_2SO_4}{1\,\%\,H_gSO_4} A \stackrel{ ext{isomerism}}{\longrightarrow} CH_3-CH_2- \stackrel{C}{\underset{||}{\longrightarrow}} -CH_3$$

. Structure of 'A' and type of isomerism in the above reaction are respectively.

A. But-1-en-2-ol, metamerism

B. But-1-cn-ol,tautomerism

C. But-2-en-2-ol,geometrical

D. But-1-en-2-ol,tautomerism.

Answer: D



3. Compound A and C in the following reaction are $CH_3CHO \xrightarrow{(i) CH_3MgBr}_{(ii) H_2O} A \xrightarrow{H_2SO_4}_{\Delta} B \xrightarrow{\text{Hydroboration}}_{\text{Oxidation}} C$ A. identical

B. position isomer

C. functional isomer

D. optical isomer

Answer: B

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4. Which is the most suitable reagent for the following conversion ?

$$CH_3-CH=CH-CH_2-\overset{O}{\overset{ert l}{C}}-CH_3
ightarrow
o
CH_3-CH=CH-CH_2-\overset{O}{\overset{ert l}{C}}-OH$$

A. Tollen's reagent

B. Benzoyl peroxide

C. I_2 and NaOH solution

D. Sn and NaOH solution

Answer: C



5. Toluene
$$\xrightarrow{KmNO_4} A \xrightarrow{SOCI_2} B \xrightarrow{H_2IPd}_{BaSO_4}$$
 the product 'C' is

A. $C_6H_5CH_2-OH$

B. C_6H_5CHO

 $\mathsf{C.}\, C_6H_5COOH$

 $\mathsf{D.}\, C_6H_5CH_3.$

Answer: B



6. The increasing order of the rate of HCN addition of compound a-d

is

(i) HCHO

(ii) CH_3COCH_3

(iii) $PhCOCH_3$

(iv) PhCOPh

A. iv < ii < iii < i

B. iv < iii < ii < i

 $\mathsf{C}.\,iii < iv < ii < i$

D. i < ii < iii < iv.

Answer: B

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7. Among the following acids, which has the lowest pK_a value?

A. CH_3COOH

 $\mathsf{B}.\,HCOOH$

 $C. (CH_3)_2 CHCOOH$

 $\mathsf{D}.\,CH_3-CH_2-OH.$

Answer: B

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8. The correct order of acidity in given compounds

(i) FCH_2COOH

(ii) $CICH_2COOH$

(iii) NO_2CH_2COOH

(iv) CH_3COOH .

A. i > ii > iii > iv

 $\mathsf{B}.\,iv>iii>ii>i$

 $\mathsf{C}.\,iii>iv>i>ii$

$$\mathsf{D}.\,iii>i>ii>iv.$$

Answer: D



$$\textbf{9.} CH_3 CHO + HCHO \xrightarrow[\text{Heat}]{\text{dil. NaOH}} A \xrightarrow[H_3O^+]{HCN} B$$

the structure of compound B is

$$\begin{split} \textbf{A}. \ CH_2 &= CH - \underset{OH}{COOH} \\ \textbf{B}. \ CH_2 &= CH - \underset{CN}{C} H - OH. \\ \vdots \\ \textbf{C}. \ CH_3 - CH_2 - \underset{OH}{C} H - COOH. \\ \textbf{D}. \ CH_3 - \underset{OH}{C} H - COOH. \end{split}$$

Answer: A

:



- A. 2-Formylhex-2-ene-3-one
- B. 5-methyl-4-oxo-hex-2-en-5-al
- C. 3-keto-2-methylhex-5-enal
- D. 3-keto-2-methylhex-4-enal.

Answer: D



X,A,Y,Z.

A. A-methoxymethane, X-ehtanol, Y-ethanoic acid, Z-semicarbazide

B. A-ethanol, X-ethano, Y-but -z-enal, Z-semicarbazone.

C. A-ehtanol,X-Acetaldehyde, Y-Butanone, Z-Hydroazone

D. A-Methoxymethane, X-ethanoicacid, Y-acetate ion and Z-Hydrazine.

Answer: B

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12. Which of the following reactions will not result in the formation of carbon- carbon bond?

A. Cannizaro's reaction

B. Wurtz reaction

C. Friedel craft reaction

D. Reimer Tiemann reaction

Answer: A

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13. Propionic acid with Br_2/P yields a dibromo product. Its structure would be

$$\begin{array}{c} \overset{Br}{\underset{Br}{\overset{|}{_{Br}}}} \\ \text{A. } CH_{3} - CH_{2} - \overset{Br}{\underset{Br}{\overset{|}{_{Br}}}} \\ \text{B. } CH_{3} - \overset{Br}{\underset{Br}{\overset{|}{_{C}}}} - CH_{2}COOH \\ \overset{B}{\underset{Br}{\overset{|}{_{Br}}}} \\ \text{C. } CH_{3} - CH_{2} - COBr \\ \overset{B}{\underset{Br}{\overset{|}{_{Br}}}} \\ \text{D. } CH_{3} - CH - COOH \\ \overset{B}{\underset{Br}{\overset{|}{_{Br}}}} \\ \overset{B}{\underset{Br}{\overset{|}{_{Br}}}} \end{array}$$

Answer: A



14. Reduction of aldehydes and ketones into hydrocarbons using Zn-Hg+HClcalled.

A. Dow process

B. Cope reduction

C. Wolf -kishner reduction

D. Clemmenson's Reduction

Answer: D



15. Acetophenone when reacted with a base C_2H_5ONa yields a stable

compound which has the structure



Answer: D



16. Which of the following compounds do not undergo aldol condensation?

A. $CH_3 - CHO$

B. $C_6H_5 - CHO$

Answer: B::D



17. Treatement of compound
$$Ph-O-\overset{O}{\overset{||}{C}}-Ph$$

with NaOH solution yields

A. Phenol

B. Sodium phenoxide

C. Sodium benzoate

D. Benzophenone

Answer: B::C

18. Which of the following conversion can be carried out by Clemmensen reduction ?

A. Benzaldehyde to benzyl alcohol

B. Cyclohexanone to cyclohexane

C. Benzoylchloride into benzaldehyde

D. Benzophenone to diphenyl methane.

Answer: B::D

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19. Through which of the following reactions number of carbon atoms

can be increased in the chain ?

A. Grignard reagent

B. Cannizaro reaction

C. Aldol condensation

D. HVZ reaction.

Answer: A::C

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20. Benzophenone can be obtained by

A. Benzoyl chloride + Benzene +anhy. $AlCI_3$.

B. Benzoyl chloride + Diphenylanion

C. Benzoyl chloride + Phenyl magnesium chloride

D. Benzene + carbon monoxide $+ ZnCI_2$.

Answer: A::B



21. Assertion (A) Formaldehyde is a planar molecule.

Reason (R) It contains sp^2 hybridised carbon atom.

A. Assertion and reason both are CORRECT and reason is the

CORRECT explanation of the assertion.

B. Assertion and reason both are wrong statements.

C. Assertion is correct statement but reason is wrong statement.

D. Assertion is wring statement but reason is correct statement.

Answer: A

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22. Assertion (A) The α -hydrogen atom in carbonyl compounds is less acidic.

Reason (R) The anion formed after the loss of α -hydrogen atom is resonance stabilised.

A. Assertion and reason both are CORRECT and reason is the

CORRECT explanation of the assertion.

B. Assertion and reason both are wrong statements.

C. Assertion is correct statement but reason is wrong statement.

D. Assertion is wring statement but reason is correct statement.

Answer: D

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23. Match the reactions given in Column I with the suitable reagents

given in Column II.

	Column 1 (Reactions)		Column il (Reagents)
A.	Benzophenone → Diphenylmethane	1.	LIAIH4
В.	Benzaldehyde → 1-phenylethanol	2.	DIBAL-H
C.	Cyclohexanone \rightarrow Cyclohexanol	3.	Zn(Hg)/Conc. HCl
D.	Phenyl benzoate→ Benzaldehyde	4.	CH ₃ MgBr

24. Match the reactions given in Column I with the suitable reagents given in Column II.





25. How many of following compound undergo aldol condensationMethanol, 2-Methylpentanal, bnzaldehyde, benzophenone

cyclohexanone, 1-phenyl propanone, butan-1-ol,2,2-Dimethylbutanal.
0 1 2 3 4 5 6 7 8 9
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Very Short Answer Type Questions 1 Mark
1. Arrange the following compounds in an increasing order of their acid
strengths :
$(CH_3)_2 CHCOOH, CH_3 CH_2 CH(Br) COOH, CH_3 CH(Br) CH_2 COOH$
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2. Draw the structure of he compound whose IUPAC name is 4-chloropentan2-one



7. Arrange the following compounds in increasing order of their boiling

points.

 $CH_3CHO, CH_3CH_2OH, CH_3OCH_3, CH_3CH_2CH_3$

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8. How is acetone obtained from ethanol ?
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9. Why do aldehydes and ketones have lower boiling point than alcohols ?
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10. Write reaction between acetyl chloride and dimethyl cadmium.

11. What happens when CH_3CHO is treated with $K_2Cr_2O_7$ in presence of H_2SO_4 ?



13. Give balanced equation and name of products when CH_3COOH is treated with PCl_5 ?

14. What product is obtained when ethyl benzene is oxidized with alkaline $KMnO_4$?



15. CH_3CHO is more reactive than CH_3COCH_3 towards reaction

with HCN. Give reason.

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16. What is RDX ?

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17. HOOC - CH = CH - COOH



Short Answer I Type Questions 2 Marks





5. An organic compound X has molecular formula $C_5H_{10}O$. It does not reduce Fehling's solution but forms a bisulphite compound. It also gives positive lodoform test. What are possible structure of X ? Explain your reasoning relating structure.





9. Which acid of each pair shown here would you expect to be stronger

 $?CH_3CO_2H$ or CH_2FCO_2H

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10. Which acid of each pair shown here would you expect to be stronger

 $? CH_2FCO_2H$ or CH_2CICO_2H

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11. Which acid of each pair shown here would you expect to be stronger

? $CH_2FCH_2CH_2CO_2H$ or $CH_3CHFCH_2CO_2H$.







16. Accout for the following : (i) Oxidation of toluence to C_6H_5CHO

with CrO_3 is carried out in presence of acetic anhydride.

17. Accout for the following : (ii) Melting point of an acid with even number is higher than those of its neighbours with odd number of carbons atoms.

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18. Distinguish between : (i) C_2H_5OH and CH_3CHO .

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19. Distinguish between : (ii) $C_6H_5COCH_3$ and $C_6H_5CH_2CHO$.



20. Complete the following reactions by identitying A:

 $A + \mathrm{Hydrogen}(g) \xrightarrow{Pd \, / \, BaSO_4} (CH_3)_2 CH - CHO$

21. Compelte the following reactions by identifying B and C: (ii) $CH_3 = \bigcup_{i=1}^{CH_3} - C - CH_3 + NaOH \rightarrow B + C.$

$$|$$
 $||$ CH_3 O

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22. Benzaldehyde gives a positive test with Tollen's reagent but not with

Fehling's and Benedict solutionws. Why?



23. Aldehydes usually do not form stable hydrates but chloral normally

exists as chloral hydrate. Give reason.

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24. Give possible explanation for the following : (i) Cyclohexanone forms cyanohydrins in good yield but 2,2,6 trimethyl-cyclohexanone does not .

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25. Give possible explanation for the following : (ii) There are two - NH_2 groups is semicarbazide . However, only one is involved in formation of semi carbonzone.

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26. Aldehydes are easily oxidisable yet propanal can conveniently br prepared by the oxidation of propanol by acid $K_2Cr_2O_7$.



27. Do the following conversions in not more than two steps. (i) Benzoic

acid to Benzalolehyde (ii) Propanone to propene.

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28. Write the reactions involved in the following reactions: (i) Clemmensen reduction
(ii) Cannizzaro reaction
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29. Convert the following (i) Ethyl benzene to benzoic acid
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30. Convert the following (ii) Ethanal to but-2-enal.

Short Answer Ii Type Questions 3 Marks

1. ILLustrate the following name reactions : Hell - Volhard Zelinsky

reaction

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2. Wolff-Kishner reduction is :

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3. ETARD REACTION



 $C_{6}H_{5}CHO \xrightarrow{H_{2}NCONHNH_{2}}$



9. Predict the organic products of the following reactions :



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10. Write chemical reaction to affect the following transformations :

Butan-1-ol \rightarrow Butanoic acid.

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11. Write chemical reaction to affect the following transformations :

Benzyl alcohol to phenynylethanoic acid

12. Write chemical reaction to affect the following transformations :

3-Nitrobromobenzene to 3-nitrobenzoic acid.

Watch Video Solution 13. Write chemical reaction to affect the following transformations : 4-Methylacetophenone to Terephthalic acid Watch Video Solution 14. Write chemical reaction to affect the following transformations :

Cyclohexene \rightarrow Hexane-1,6 dioic acid



15. Write chemical reaction to affect the following transformations :

Butanal \rightarrow Butanoic acid.





16. Draw the structure of the following derivatives : 2,4- dinitrophenylhydrazone of C_6H_5CHO

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17. Draw the structure of the following derivatives : Cyclopropanone oxime

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18. Draw the structure of the following derivatives :Acetaldehydedimethylacetal

19. Draw the structure of the following derivatives : Semicarbazone to

cyclobutanone

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20. Draw the structure of the following derivatives : Ehylene ketal of

hexan-3-one

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21. Draw the structure of the following derivatives : Methylhemiacetal of

formaldehyde.



22. Draw the structure of a carbonyl group and indicate :

hybridized state of carbon .



25. Complete the following as missing starting material , reagent or products :

(i)
$$\cdots \cdots \xrightarrow{O_3} 2 = O$$



28. How can the following converted :

(i) Ethanol \rightarrow Acetone



29. How will you convert Benzene to acetophenone?

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30. How will you convert benzoic acid into benzaldehyde ?

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31. Give reasons for the following :

(i) Carboxylic acids do not give characteristic reactions of carbonyl

group.

32. Give reasons for the following :

(ii) Treatment of C_6H_5CHO with HCN gives a mixture of two isomers

which cannot be separted even by fractional distillation.

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33. Give reasons for the following :

(iii) Sodium bisulphite is used for purification of ketones and aldehydes.

34. <i>CH</i> ₃	Give GCHO	the and C	chemical $_{6}h_{5}CHO$.	test	of	distinguish	between	:
C	Watch	n Videc	Solution					

35. Write tests of distinguish between :

(ii) $C_6H_5 - OH$ and CH_3COOH .



(i) Benzaldehyde to acetophenone



38. Convert :

(ii) Malonic acid to acetic acid



40. Write the structues of organic compound A to F in the following sequence of reactions :



41. Complete the following :

(i) $CH_{3}CONH_{2} + HNO_{2}
ightarrow$



42. Complete the following :

(ii) $CH_{3}CONH_{2}+NaOH+Br_{2}
ightarrow$

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43. Complete the following :



44. Write the structures of A,B,C,D and E in the following reactions :



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Long Answer Type Questions 5 Marks

1. Which of the following compounds would undergo Aldol condensation, which the Cannizzaro reaction and which neither ? Write the structures of the expected produces of aldol condensation and Cannizzaro reaction :

(i) Methanal (ii) 2-Methylpentanal (iii) Benzaldehyde (iv) Benzophenone
(v) Cyclohexanone (vi) 1-Phenylpropanone (vii) Phenylacetaldehyde (viii)
Butan-1-ol (ix) 2,2 Dimethylbutanal.

2. An organic compound 'A' (C_3H_6O) is resistant to oxidation but forms compound 'B' (C_3H_8O) on reduction. 'B' reacts with HBr to form the compound 'C' .'C' with Mg forms Grignard's reagent 'D' which reacts with 'A' to form a product which on hydrolysis gives 'E'. Identify 'A' to 'E'.



4. Give simple tests to distinguish between the following pair of compounds.

(i) Benzaldehyde and Benzoic acid

(ii) Propanal and propanone.

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5. (i) How will you prepare (a) acetic anhydride and (b) acetyl chloride

from CH_3COOH ? Write the equation involved in each case.

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6. (ii) Why is the boiling point of acid anhydride higher than the acid

from which it is obtained ?



10. Complete the following reactions and write main products : (iv)

$$CH_3COOH + NH_3 \xrightarrow{\Delta}$$
.



13. Complete the following reactions and write main products : (vii) $CH_3CHO \xrightarrow{LiAIH_4}$

14. Complete the following reactions and write main products : (viii)

(viii) СН₃СНО | –

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15. Complete the following reactions and write main products : (ix)

 $CH_3COR + NaOH
ightarrow$.

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16. Complete the following reactions and write main products : (x)

 $C_6H_5CHO+NH_2-NH_2
ightarrow$

17. Give reasons for the following : (i) C_6H_5COOH is weaker than
formic acid.
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18. Give reasons for the following : (ii) HCOOH and CH_3COOH
differentiated by Tollen's reagent.
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19. Give reasons for the following : (iii) R - COOH do not give characteristic reaction with > C = O.

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20. Why is carboxylic acid a stronger acid than phenol?

21. Give reasons for the following : (v) Acid amides are weakly basic in

nature .